

Cartland in The Upjohn Company Laboratories.<sup>3</sup> It was quite surprising to find that none of these ureas were effective to any great degree when injected intraperitoneally into white rats although, in a second paper, it will be shown that some of the corresponding acetamides are strong hypnotics.

The ureas were obtained by the following general method: conversion of malonic ester into the required disubstituted derivative,  $R_2C(COOC_2H_5)_2$ , hydrolysis of the ester, elimination of carbon dioxide with the formation of a disubstituted acetic acid,  $R_2CH-COOH$ , preparation of the corresponding acetyl chloride and treatment of the latter with urea or methylurea.<sup>4</sup>

The new intermediates are described in Table I.

### Experimental Part

The procedures are illustrated in the case of allyl- $\beta$ -phenylethylacetylurea.

To sodium ethylate, prepared from 1000 cc. of absolute alcohol and 46 g. of sodium, there was added, slowly, 320 g. of malonic ester. After one hour the mixture was stirred and refluxed and 370 g. of  $\beta$ -phenylethyl bromide added during the course of two hours. The mixture was stirred and refluxed for two hours longer, most of the alcohol removed by distillation and 500 cc. of water added to the cold residue. The ester layer was separated and the aqueous layer extracted six times with 75-cc. portions of ether; yield of  $\beta$ -phenylethylmalonic ester 424 g. or 80% of the calcd. amount; b. p. 182-185° (12 mm.).<sup>5</sup>

(3) Their results will be published by them in detail in another journal.

(4) Davis and Blanchard, *THIS JOURNAL*, **51**, 1797 (1929).

(5) Dolique [*Ann. chim.*, [10] **15**, 447 (1931)] reported 184-185° at 15 mm.

In a manner similar to that described above, allyl- $\beta$ -phenylethylmalonic ester was prepared from sodium ethylate, obtained from 300 cc. of alcohol and 13.8 g. of sodium, 159 g. of  $\beta$ -phenylethylmalonic ester and 73.2 g. of allyl bromide; yield 159 g. or 88% of the calcd. amount.

A mixture of 80 g. of potassium hydroxide, 250 cc. of 70% alcohol and 101 g. of the disubstituted malonic ester was refluxed for twenty-four hours, most of the alcohol removed, 250 cc. of water added and then about 140 cc. of hydrochloric acid added to the cold mixture. The crude acid separated from the acidic mixture as an oil but soon solidified; yield 78 g. or 94% of the calcd. amount.

Fifteen grams of the allyl- $\beta$ -phenylethylmalonic acid was heated in an oil-bath at 180° until most of the carbon dioxide had been evolved and then heated at 150-160° for six hours; yield 10.7 g. or 86% of the calcd. amount.

A mixture of 28 g. of the acetic acid and 45 g. of commercial thionyl chloride was heated for two hours on a steam-bath after the initial vigorous reaction had subsided and the excess thionyl chloride then removed; the yield of acid chloride was 21 g. or 70% of the calcd. amount.

Nine grams of allyl- $\beta$ -phenylethylacetyl chloride and 9.6 g. of dry urea were heated in an oil-bath at 125°. As soon as the material had melted the temperature was dropped to 110-115°. After six hours the product was cooled and rubbed in a mortar with enough 10% sodium carbonate solution to keep the mixture alkaline. The solid material was filtered, washed with water and recrystallized; yield 8.9 g. or 90% of the calcd. amount.

### Summary

A number of new disubstituted malonic esters, malonic acids, acetic acids, acetyl chlorides and acetylureas have been described.

No strong hypnotics were found among the disubstituted acetylureas,  $R_2CH-CO-NH-CO-NH_2$ , studied.

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## Acid Amides as Hypnotics. II. Acetamides<sup>1</sup>

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It has been known for a long time that certain representatives of three types of acid amides—substituted acetamides, acylureas and cyclic amides such as substituted barbituric acids and hydantoins—exhibit strong hypnotic activity.

During late years, due to the popularity of barbituric acid compounds, the study of the acetamide type has been neglected except for

(1) This paper represents part of a dissertation submitted to the Horace H. Rackham School of Graduate Studies by A. P. Centolella in partial fulfillment of the requirements for the degree of Doctor of Philosophy in the University of Michigan.

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the publications of Volwiler and Tabern<sup>3</sup> and of Junkmann.<sup>4</sup>

Most of the acetamides, described hitherto as compounds which possess hypnotic action, are trisubstituted derivatives such as diethylallylacetamide. Since, in general, trisubstituted acetamides are more difficult to obtain than the disubstituted compounds we have prepared a considerable number of the latter in order to determine their activity as hypnotics.

(3) Volwiler and Tabern, *THIS JOURNAL*, **58**, 1353 (1936).

(4) Junkmann, *Arch. expil. Path. Pharmacol.*, **186**, 552 (1937).



TABLE II  
 SUBSTITUTED ACETAMIDES

The amides were recrystallized in the following manner: compound 2 from water; compounds 8, 9, 10, 11, 12, 15, 16, 20, 22, 24, 25, 26 and 27 from petroleum ether (90–100°); compounds 18, 19 and 21 from dilute alcohol; compound 17 from acetone; compound 13 from 50% acetone; compounds 5 and 23 from a mixture of one part acetone and two parts petroleum ether (90–100°); compounds 1, 4, 6, 7, 10 and 14 from a mixture of one part acetone and nine parts petroleum ether.

| Acetamide                                   | M. p., °C.           | Formula  | Nitrogen, % |       |
|---|----------------------|--|-------------|-------|
|   |                      |  | Calcd.      | Found |
| 1 Ethylbenzyl                               | 117–118              | C <sub>11</sub> H <sub>15</sub> ON               | 7.90        | 8.00  |
| 2 Ethyl-β-phenylethyl                       | 105–106 <sup>a</sup> | C <sub>12</sub> H <sub>17</sub> ON               | 7.32        | 7.30  |
| 3 N-Methylethyl-β-phenylethyl               | 98–99                | C <sub>13</sub> H <sub>19</sub> ON               | 6.83        | 6.93  |
| 4 N-Ethylethyl-β-phenylethyl                | 72–73                | C <sub>14</sub> H <sub>21</sub> ON               | 6.39        | 6.56  |
| 5 Di-β-phenylethyl                          | 162–163              | C <sub>18</sub> H <sub>21</sub> ON               | 5.53        | 5.42  |
| 6 N-Methyl-di-β-phenylethyl                 | 124–125              | C <sub>19</sub> H <sub>23</sub> ON               | 5.24        | 5.15  |
| 7 N-Butyl-di-β-phenylethyl                  | 86–87                | C <sub>22</sub> H <sub>29</sub> ON               | 4.43        | 4.29  |
| 8 β-Cyclohexylethyl-β'-phenylethyl          | 170–171              | C <sub>18</sub> H <sub>27</sub> ON               | 5.14        | 5.41  |
| 9 Propyl-β-phenylethyl                      | 109–110              | C <sub>13</sub> H <sub>19</sub> ON               | 6.81        | 6.78  |
| 10 N-Methylpropyl-β-phenylethyl             | 93–94                | C <sub>14</sub> H <sub>21</sub> ON               | 6.39        | 6.33  |
| 11 N-Ethylpropyl-β-phenylethyl              | 84–85                | C <sub>15</sub> H <sub>23</sub> ON               | 6.00        | 6.08  |
| 12 N-β-Hydroxyethylpropyl-β'-phenylethyl    | 74–75                | C <sub>15</sub> H <sub>23</sub> O <sub>2</sub> N | 5.63        | 5.70  |
| 13 N-Butylpropyl-β-phenylethyl              | 69–70                | C <sub>17</sub> H <sub>27</sub> ON               | 5.36        | 5.23  |
| 14 Isopropyl-β-phenylethyl                  | 121–122              | C <sub>13</sub> H <sub>19</sub> ON               | 6.81        | 6.85  |
| 15 N-β-Hydroxyethylisopropyl-β'-phenylethyl | 83–84                | C <sub>15</sub> H <sub>23</sub> O <sub>2</sub> N | 5.63        | 5.55  |
| 16 Allyl-β-phenylethyl                      | 90–91                | C <sub>13</sub> H <sub>17</sub> ON               | 6.89        | 7.13  |
| 17 Butyl-β-phenylethyl                      | 124–125              | C <sub>14</sub> H <sub>21</sub> ON               | 6.40        | 6.56  |
| 18 N-Methylbutyl-β-phenylethyl              | 108–109              | C <sub>15</sub> H <sub>23</sub> ON               | 6.00        | 5.75  |
| 19 N-Ethylbutyl-β-phenylethyl               | 71–72                | C <sub>16</sub> H <sub>25</sub> ON               | 5.67        | 5.85  |
| 20 N-β-Hydroxyethylbutyl-β'-phenylethyl     | 66–67                | C <sub>16</sub> H <sub>25</sub> O <sub>2</sub> N | 5.36        | 5.25  |
| 21 N-Butylbutyl-β-phenylethyl               | 59–60                | C <sub>18</sub> H <sub>29</sub> ON               | 5.28        | 5.01  |
| 22 Isobutyl-β-phenylethyl                   | 89–90                | C <sub>14</sub> H <sub>21</sub> ON               | 6.40        | 6.61  |
| 23 Ethyl-γ-phenylpropyl                     | 118–119              | C <sub>13</sub> H <sub>19</sub> ON               | 6.81        | 6.82  |
| 24 Ethyl-δ-phenylbutyl                      | 107–108              | C <sub>14</sub> H <sub>21</sub> ON               | 6.40        | 6.43  |
| 25 Ethyl-ε-phenylamyl                       | 98–99                | C <sub>15</sub> H <sub>23</sub> ON               | 6.01        | 6.12  |
| 26 Ethyl-ζ-phenylhexyl                      | 113–114              | C <sub>16</sub> H <sub>25</sub> ON               | 5.67        | 5.75  |
| 27 Ethylcinnamyl                            | 94–96                | C <sub>13</sub> H <sub>17</sub> ON               | 6.89        | 6.72  |

<sup>a</sup> Levy [*Compt. rend.*, **194**, 175 (1932)] found 104° but did not record an analysis.

troduced into the amide group; R' = methyl, ethyl, butyl or β-hydroxyethyl. Three compounds in this series proved to be strong hypnotics: N-ethylethyl-β-phenylethylacetamide, N-ethylpropyl-β-phenylethylacetamide and N-β-hydroxyethylpropyl-β'-phenylethylacetamide.

Six compounds of the general formula III were prepared in which  $x = 1, 2, 3, 4, 5$  and 6. Two of these compounds are strong hypnotics, namely, ethyl-β-phenylethylacetamide and ethyl-ε-phenylamylacetamide.

### Experimental Part

The amides were obtained by interaction of the disubstituted acetyl chloride with ammonia or the required amine. The new acid chlorides and their general method of preparation were described previously.<sup>5</sup>

Allyl-β-phenylethylacetamide was obtained in 87% yield when 4.5 g. of allyl-β-phenylethylacetyl chloride was added slowly, with vigorous stirring, to 20 cc. of ammonia water which was cooled with ice.

To prepare N,N'-bis-(diethylacetyl)-ethylenediamine

20 cc. of a 66% aqueous solution of ethylenediamine was stirred, cooled and 2.6 g. of diethylacetyl chloride added dropwise. The precipitate was filtered, washed with water, dried and recrystallized four times from petroleum ether; yield 2.4 g. or 95% of the calcd. amount.

No attempt was made to discover a satisfactory procedure for the preparation of diethylthioacetamide. However, sufficient material for our purpose was obtained when 22 g. of phosphorus pentasulfide and 23 g. of diethylacetamide were mixed intimately, 175 cc. of benzene added and the mixture refluxed for five minutes. The warm benzene was decanted from the pasty residue and the latter extracted with three 25-cc. portions of hot benzene. The solvent was removed from the combined benzene solutions and the gummy, yellow residue recrystallized from petroleum ether (90–100°).

### Summary

A number of new dialkyl-, alkyl-β-cyclohexylethyl- and alkyl-β-phenylethylacetamides have been described. In some instances N-alkyl derivatives were prepared.

Some of the amides have been found to be strong hypnotics.

(5) Blicke and Centolella, *This Journal*, **60**, 2923 (1938).